

S0040-4039(96)00195-5

Intramolecular [2+2] Photocycloaddition - Cyclobutane Rearrangement. A Novel Stereocontrolled Approach to Highly Substituted Cyclopentanones#

Subrata Ghosh*, Debasis Patra and Susanta Samajdar

Department of Organic Chemistry, Indian Association for the Cultivation of Science, Jadavpur, Calcutta - 700 032, India

Abstract: A simple stereocontrolled route involving intramolecular [2+2] photocycloaddition followed by rearrangement of the resulting cyclobutane derivatives is described for the construction of cyclopentanones with substituents upto three contiguous chiral centres.

The synthetic and stereochemical problem associated with complex cyclopentanoid natural products provide ample scope for development of new methods¹ for the construction of cyclopentane rings. Many elegant strategies have been developed for the synthesis of vicinally substituted cyclopentanes². However, direct methods of constructing cyclopentanes³ of the general structure 1 with substituents on more than two contiguous carbon centres are scarce, despite the presence of these structural units in a wide range of natural

products e.g., dictymal 2,⁴ clavularane 3,⁵ ceroplastol 4,⁶ etc. Herein, we report a direct stereocontrolled approach for the construction of cyclopentanones with substituents on three contiguous centres.

The key concept lies in the cuprous triflate (CuOTf) catalysed photocycloaddition of the diene 8 to form the cyclobutane derivative 9 followed by its stereospecific rearrangement involving central bond migration to the cyclopentanones 10 (Scheme 1).² It is the CuOTf catalysed cycloaddition that plays the pivotal role in determining the stereochemistry of the resulting cyclopentanones. As CuOTf catalyses E,Z-olefin isomerisation during irradiation, photocycloaddition of dienes e.g. 8 without the C₃ - substituents normally ⁸ produces a

[#]Dedicated to Professor U. R. Ghatak on the occasion of his 65th birth anniversary

mixture of adducts arising from both E and Z isomers. We anticipate that with dienes 8 having the C_3 -substituents, the Cu(I)-diene complex N 7 formed from Z isomer prior to photocycloaddition gets destabilised due to steric crowding between Me and R 2 inhibiting its formation. Furthermore, of the two C_3 -substituents the bulkier one prefers to occupy the exo position 2 in the Cu-complex. Photocycloaddition of the diene 8 is thus expected to produce the exo adduct 9 as the major product arising through the Cu - complex X derived from the E - isomer.

The dienes 8 required for this investigation were prepared from the ketones 5 on reaction with ethoxy vinyl lithium followed by coupling of the resulting carbinols 6 with the bromides 7. Irradiation of an ether solution of the diene 8a prepared from acetone, in the presence of CuOTf afforded a chromatographically inseparable mixture of the exo 8ax and its corresponding endo cyclobutane derivative in ca. 2.5:1 ratio (GC). Treatment of this mixture with triflic acid (TfOH) effected smooth rearrangement to afford the cyclopentanone 10a and its C₄-epimer in a ratio nearly identical to that of the starting cyclobutanes. However, with the diene 8b having a Ph at C₇, photocycloaddition proceeded, to our expectation, with complete stereocontrol to produce exclusively a single adduct 8bx⁹ (¹H and ¹³C) (53%). The exo stereochemical assignment to the adduct follows from its rearrangement to produce the single cyclopentanone derivative 10b in 55% isolated yield. The trans relationship between the C₃-and C₄-hydrogens is established by the coupling constant (J=11.6 Hz), closely comparable to the coupling constant reported ^{3b} for 1,2- disubstituted cyclopentanes. With the establishment of the structures of the cyclopentanone derivative 10b and the cyclobutane derivative 9ax, the major photoadduct and the major cyclopentanone obtained from the diene 8a was assigned respectively the structures 9ax and 10a.

The general applicability and synthetic potential of this protocol is established by the synthesis of the cyclopentanones 10c and 10d as described below. The triene 8c on irradiation in presence of CuOTf afforded a mixture of the exo cyclobutane derivative 9cx and its corresponding endo isomer in ca. 2.2: 1 ratio. Before carrying out the rearrangement, the double bond in the mixture of the photoadducts was oxidatively cleaved and the resulting acid mixture after esterification was equilibriated with refluxing NaOMe - MeOH to afford the thermodynamically more stable⁸ exo cyclobutane derivative 11 as the major product (58%). Rearrangement of the cyclobutane derivative 11 afforded the cyclopentanone 10c (76%). On the contrary the triene 8d afforded a single photoadduct¹⁰ (41%) to which the exo structure 9dx was assigned based on its transformation to the cyclopentanone derivative 10d through the cyclobutane ester 12. The coupling constants between the C₃,

Scheme 1 Reagents and Conditions: I, Bu'Li, ethyl vinyl ether, THF, -70°C to rt, 80-95%; ii, NaH, THF, HMPA, reflux, 62-92%; iii, hv, Et₂O, CuOTf, 40-50%; iv, TfOH, CH₂Cl₂, -78°C to rt. 52-76%; v, OsO₄, NaIO₄, Et₂O-H₂O; Jones oxidation, O°C; CH₂N₂, Et₂O; NaOMe-MeOH, reflux, 54 - 58%.

C₄-hydrogens of the cyclopentanones 10c and 10d respectively(10.6 and 11.0 Hz) dictates the trans stereochemical assignment of the substituents at these centres which in turn confirms the exo structures 9cx and 9dx for the cyclopentane derivatives. The resistance of the cyclopentanone ester 10d to undergo epimerisation confirms this stereochemical assignment. The cyclopentanone derivative 10c is suitably functionalised for elaboration to the gem-dimethylated natural products e.g. capnellene while the cyclopentanone derivative 10d, because of its structural and stereochemical similarity, offers possibility for the synthesis of the terpenes 2 - 4.

This investigation thus provides with an expedient stereocontrolled route to construct at the carbonyl carbon of appropriately chosen acyclic ketones, cyclopentanones with substituents upto three contiguous chiral centres.

Acknowledgement: Financial support from the Department of Science and Technology, New Delhi is gratefully acknowledged.

REFERENCES AND NOTES

- 1. a) Paquette, L. A. Top. Curr. Chem. 1984, 119, 1; b) Ramaiah, M., Synthesis 1984, 529.
- 2. Patra, D. and Ghosh, S., J. Org. Chem. 1995, 60, 2526 and references cited therein.

- a) Saito, S.; Hirohara, Y.; Narahara, O.; Moriwake, T., J. Am. Chem. Soc. 1989, 111, 4533; b) Llera, J. M.; Fraser-Reid, B. J. Org. Chem. 1989, 54, 5544; c) Curran, D. P.; Abraham, A. C.; Liu, H. J. Org. Chem. 1991, 56, 4335; d) Takacs, J. M.; Myoung, Y. C. Tetrahedron Lett. 1992, 33, 317.
- 4. Segawa, M.; Enoki, N.; Ikura, M.; Hikichi, K.; Ishida, R.; Shirahama, H.; Matsumoto, T. Tetrahedron Lett. 1987, 28, 3703.
- 5. Braekman, J. C.; Daloze, D.; Schubert, R.; Albericci, M.; Tursch, B.; Karlsson, R. Tetrahedron 1978, 34, 1551
- 6. Rios, T.; Colunga, F. Chem. Ind. 1965, 1184.
- 7. Ghosh, S.; Raychowdhuri, S. R.; Salomon, R. G. J. Org. Chem. 1987, 52, 83.
- 8. Salomon, R. G.; Coughlin, D. J.; Ghosh, S.; Zagorski, M. J. Am. Chem. Soc. 1982, 104, 998.
- Selected spectral data: 9bx, NMR: ¹H(200 MHz) 1.21(t, J=7Hz, 3H, 1.24(s, 3H), 1.40(s, 3H), 2.18-9. 2.28(m, 1H), 2.68-2.97(m, 3H), 3.29-3.48(m, 2H), 3.68(d, J=9.6 Hz, 1H), 3.92(dd, J=4.8, 9.6 Hz, 1H) and 7.12-7.33(m, 5H); ¹³C(DEPT) 15.7(Me), 21.3(Me), 24.7(Me), 32.1(CH₂), 36.1(CH), 52.9(CH), 59.2 (CH₂), 68.5(CH₂), 82.2, 85.1, 125.9, 126.6, 128.4, 129.1 and 145.5; 10b, IR(neat): 1735, 3440 cm⁻¹; NMR: ¹H (200 MHz) 1.07(s, 3H), 1.25(s, 3H), 2.23 (partly resolved dt, J=6, 11.8 Hz, C₃-H), 2.42(dd, J=11.6, 18.8 Hz, 1H), 2.80(dd, J=8, 18.8 Hz, 1H), 3.07(dt, J=8, 11.6 Hz, C₄-H), 3.7(d, J=6Hz, 18.8 Hz, 18.82H) and 6.41-7.55(m, 5H); 12, IR(neat): 1735 cm⁻¹; NMR: ¹H(200MHz) 1.11(s, 3H), 1.24(t, J=7Hz, 3H), 2.43-2.63(m, 4H), 3.02-3.18 (m, 2H), 3.56(q, J=7Hz, 2H), 3.72(s, merged under a m at 3.64-3.77, total 4H), 4.13 (dd, J=5, 10Hz, 1H) and 7.17-7.31(m, 5H); ¹³C(DEPT) 15.7(Me), 20.4(Me), 27.36(CH₂), 34.9 (CH), 36.9(CH₂), 47.9(CH), 51.9(OMe), 60.2(CH₂), 68.5(CH₂), 84.1, 87.0, 126.0, 127.8, 130.7, 138.4 and 175.1; 10d, IR(neat): 1735, 3480 cm⁻¹; NMR: ¹H(200 MHz) 1.03(s, 3H), 2.23(dd, J=18.8, 1H, 2.41-2.54(m, 1H), 2.58-2.71(m, 2H), 2.85(dt, J=7.7, 11 Hz, 1H), 3.09(d, J=13.6 Hz, 1H), 3.68(s merged with a m at 3.74, 5H) and 7.08-7.31 (m, 5H); ¹³C(DEPT) 18.7(Me), 41.38 (CH₂), 41.45(CH), 42.4(CH₂), 47.0 (CH), 52.2(OMe), 53.1, 62.1(CH₂), 126.6, 128.3, 130.3, 137.1, 175.1 and 218.4; EIMS: m/z 276(M⁺), 258, 245, 199, 172, 145, 129, 114, 91(100%), 55 and 43.
- 10. The formation of a single photoadduct is possibly the result of destabilisation of the complex N by greater gem-dialkyl effect ¹¹ of the C₃-substitutents (Me, CH₂Ph in 8d vs. Me, Me in 8a-c) that pushes away the C₃-Me towards R².
- cf. a) Jung, M. E.; Trifunovich, I. D.; Lensen, N. Tetrahadron Lett. 1992, 33, 6719; b) Jung, M. E.;
 Gervay, J. J. Am. Chem. Soc. 1991, 113, 224.

(Received in UK 23 November 1995; revised 29 January 1996; accepted 2 February 1996)